

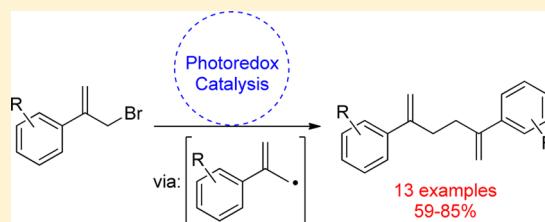
Synthesis of 2,5-Diaryl-1,5-dienes from Allylic Bromides Using Visible-Light Photoredox Catalysis

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S Supporting Information

ABSTRACT: Visible-light photoreductive coupling of 2-arylallyl bromides in the presence of the photocatalyst $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$, a Hantzsch ester, and *i*-Pr₂NEt gives 2,5-diaryl-1,5-dienes in high yield. This method avoids the use of stoichiometric metal reductants and is compatible with the presence of halogen, alkyl, electron-donating, and electron-withdrawing substituents on the aromatic ring.



INTRODUCTION

The importance of visible-light photocatalysis in organic synthesis has increased substantially in recent years, as it is a sustainable and green method and offers unique opportunities for controlling selectivity.^{1,2} In the area of free radical reactions, visible-light photocatalysis is allowing a variety of such processes to be accomplished without the use of stoichiometric adjuvants such as tin reagents. A variety of precursors such as halides,^{1d,f,3} selenosulfonates,⁴ sulfonium⁵ and sulfonyl⁶ derivatives, diazonium salts,⁷ carboxylic acids,⁸ (*N*-acyloxy)-phthalimides,⁹ *N*-phthalimidoyl oxalates,¹⁰ enamines,¹¹ and other α -amino substituted compounds¹² can be used to generate carbon radicals under these conditions. In a recent study of the reaction of tertiary radicals generated from *N*-(acyloxy)phthalimides under visible-light photoredox conditions with allylic halides as acceptors,^{9b} we observed the formation of substantial amounts of 2,5-diphenylhexa-1,5-diene as a side product in attempted couplings with α -(bromomethyl)styrene. Inasmuch as the formation of 1,5-dienes by reductive coupling of allylic halides is typically accomplished using stoichiometric metal reductants (Wurtz couplings),¹³ and no fully satisfactory method appears to be available for the synthesis of 2,5-diaryl-1,5-dienes,¹⁴ we decided to explore the utility of visible-light photoredox catalysis for preparing such dienes. The outcome of these investigations, which led to a general, high-yielding method for preparing 2,5-diaryl-1,5-dienes by utilizing allylic bromides as radical precursors for the first time in photoredox catalysis, is reported herein.

RESULTS AND DISCUSSION

Salient results of our initial optimization of the synthesis of 2,5-diphenylhexa-1,5-diene (**1a**) from α -(bromomethyl)styrene (**2a**) are summarized in Table 1. By simple omission of the *N*-(acyloxy)phthalimide radical precursor, diene **1a** was obtained in 78% yield under the conditions employed in our earlier cross-coupling studies (entry 1).^{9b} Control experiments showed that the photocatalyst, $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$, light, and

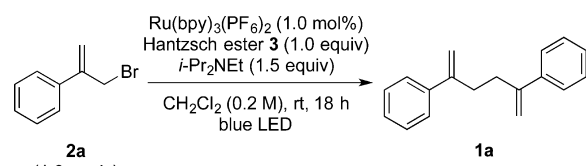
Hünig's base (*i*-Pr₂NEt) are essential (entries 2–4). In the absence of the Hantzsch ester (diethyl 1,4-dihydro-2,6-dimethylpyridine-3,5-dicarboxylate, **3**), diene **1a** was formed in low yield (24%), with 62% of bromide **2a** being recovered (entry 5). Subsequent optimization reactions revealed that the product yield and conversion was lower when the catalyst loading was decreased to 0.5 mol % and unchanged when increased from 1.0 to 1.5 mol % (entries 6 and 7). One equivalent of Hantzsch ester **3** and 2 equivalents of Hünig's base appeared optimal (entries 8–12). Dichloromethane was preferred over MeCN or THF as the reaction solvent, and increasing the starting concentration of bromide **2a** to 0.4 M was also beneficial (entries 13–16). The reaction was nearly complete after 6 h at room temperature with conversion not increasing further after 18 h (entries 17–19). Finally, combining the optimum reaction parameters led to full conversion of allylic bromide **2a** and clean formation of 2,5-diphenylhexa-1,5-diene (**1a**) in about 80% yield at scales up to 2 mmol (entry 20).

To investigate the scope of this synthesis of 2,5-diaryl-1,5-dienes, a broad selection of α -(bromomethyl)styrenes **2** was prepared from commercially available acetophenones by the two-step sequence illustrated in Scheme 1. Whereas the initial Wittig methylenation was easily accomplished,¹⁵ the allylic bromination step required some optimization (see the Supporting Information for details).¹⁶

With a reliable access to various substituted α -(bromomethyl)styrenes **2** in hand, the scope of the visible-light photoredox catalyzed formation of the 2,5-diaryl-1,5-dienes **1** was explored (Table 2). Substrates containing halogen substituents at the *meta* or *para* position gave diene products **1b–1d** in yields in excess of 80% (entries 2–4), comparable to the yield realized with the unsubstituted precursor (entry 1). A fluoro substituent at the *ortho* position prevented full conversion of the allylic bromide, resulting in a 66% yield of

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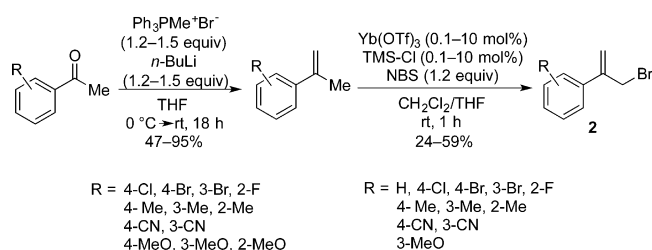
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Table 1. Optimization and Control Experiments for the Coupling of α -(Bromomethyl)styrene (2a) To Form 2,5-Diphenylhexa-1,5-diene (1a)


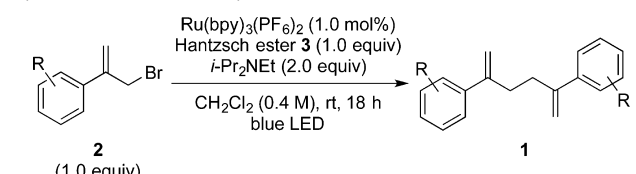
entry	modification	yield of 1a (%) ^a	recovery of 2a (%) ^a
1	none	78	4
2	no Ru(bpy) ₃ (PF ₆) ₂	ND	92
3	no light	ND	65
4	no <i>i</i> -Pr ₂ NEt	6	69
5	no Hantzsch ester 3	24	62
6	Ru(bpy) ₃ (PF ₆) ₂ (0.5 mol %)	47	35
7	Ru(bpy) ₃ (PF ₆) ₂ (1.5 mol %)	79	< 2
8	Hantzsch ester 3 (0.5 equiv)	57	25
9	Hantzsch ester 3 (0.75 equiv)	71	11
10	<i>i</i> -Pr ₂ NEt (0.5 equiv)	43	35
11	<i>i</i> -Pr ₂ NEt (1.0 equiv)	73	7
12	<i>i</i> -Pr ₂ NEt (2.0 equiv)	82	2
13	MeCN (0.2 M)	61	28
14	THF (0.2 M)	52	38
15	CH ₂ Cl ₂ (0.1 M)	69	16
16	CH ₂ Cl ₂ (0.4 M)	80	ND
17	2 h	48	43
18	6 h	72	13
19	24 h	76	6
20	<i>i</i> -Pr ₂ NEt (2.0 equiv) CH ₂ Cl ₂ (0.4 M)	81, 80 ^b , 78 ^c	ND

^aIsolated yield of 1a and recovery of 2a after silica gel chromatography, 0.2 mmol scale. ^b1.0 mmol scale. ^c2.0 mmol scale. ND = not detected.

Scheme 1. Synthesis of α -(Bromomethyl)styrenes 2 from Acetophenones



1e (entry 5). In this case, increasing the reaction time led to higher conversions of 2e; however, the yields of 1e remained in the same range. Similar results were observed in the synthesis of dienes 1f–1h containing tolyl substituents, with *ortho* substitution leading to lower conversion and lower yield (entries 6–8). Both strong electron-donating (OMe) and electron-withdrawing (CN) substituents were tolerated, giving the corresponding diene products 1i–1m in yields of 60–70% (entries 9–13). The presence of an *ortho* substituent again resulted in much lower conversion and yield of diene product 1k. However, in this case, increasing the catalyst loading to 1.5 mol % and the reaction time to 48 h resulted in diene 1k being

Table 2. Scope of the Visible-Light Photoredox Catalyzed Synthesis of 2,5-Diaryl-1,5-dienes 1


entry	R	product (yield in %) ^a
1	H (2a)	1a (81, 80 ^b)
2	4-Cl (2b)	1b (80)
3	4-Br (2c)	1c (82)
4	3-Br (2d)	1d (84, 79 ^b)
5	2-F (2e)	1e (66) ^c
6	4-Me (2f)	1f (79) ^d
7	3-Me (2g)	1g (85) ^d
8	2-Me (2h)	1h (71 ^e , 69 ^f)
9	4-OMe (2i)	1i (59)
10	3-OMe (2j)	1j (72)
11	2-OMe (2k)	1k (67) ^g
12	4-CN (2l)	1l (70)
13	3-CN (2m)	1m (70) ^d

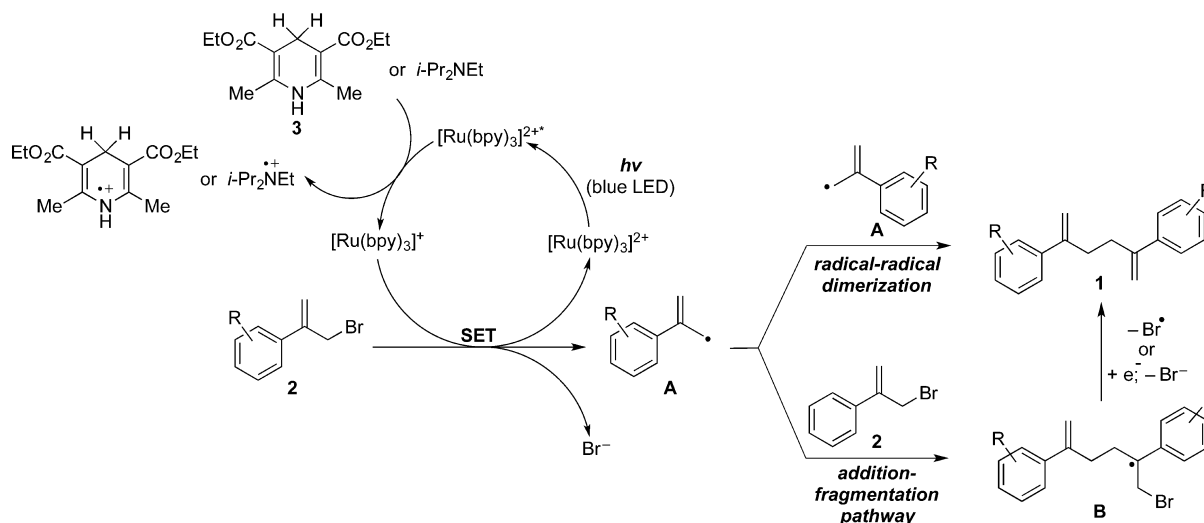
^aIsolated yield of 1 after silica gel chromatography (average of two experiments). ^b1.0 mmol scale. ^c26% recovered SM 2e (89% brsm). ^dCH₂Cl₂ (0.2 M). ^e29% recovered SM 2h (quant. brsm). ^f1.0 mmol scale, 25% recovered SM 2h (91% brsm). ^gRu(bpy)₃(PF₆)₂ (1.5 mol %), 42 h reaction time, 18% recovered SM 2k (81% brsm).

formed in 67% yield (entry 11). Substrates containing nitro substituents were partially consumed under our standard reaction conditions; however, no 1,5-diene products were formed.¹⁷ Overall, this visible-light photoredox catalyzed Wurtz-type coupling offers—in contrast to other literature-known methods¹⁴—a mild, catalytic and broadly tolerant access to 2,5-diarylhexa-1,5-dienes from easily accessible starting materials in one step and high yields.

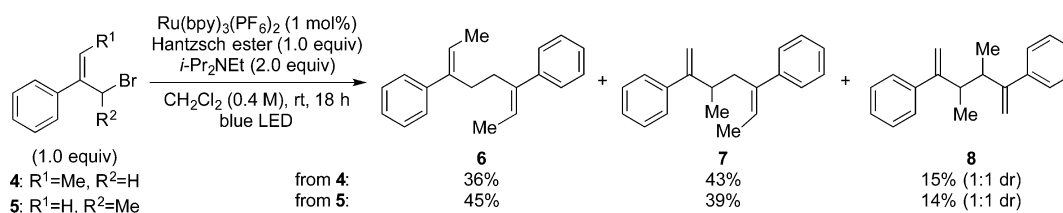
The likely mechanism for the formation of 2,5-diaryl-1,5-dienes under the conditions we report is summarized in Scheme 2.¹⁸ After visible-light photoexcitation of the catalyst Ru(bpy)₃²⁺, the corresponding excited state Ru(bpy)₃^{2+*} is quenched by a stoichiometric reductant (Hantzsch ester 3 or *i*-Pr₂NEt). Single-electron transfer from the resulting Ru(I)-species to the allylic bromide 2 generates allylic radical A with loss of bromide anion. Dimerization of A would yield the 2,5-diaryl-1,5-diene product 1. Alternatively, addition of allylic radical A to allylic bromide 2, followed by loss of bromide radical from B, or single-electron reduction of B, followed by expulsion of bromide anion, would yield product 1.^{9b}

To explore whether product formation occurs preferentially via radical dimerization or an addition–fragmentation mechanism, we examined the reaction of the unsymmetrical allylic precursors (1-bromobut-2-en-2-yl)benzene (4) and (3-bromobut-1-en-2-yl)benzene (5) (Scheme 3). Both allylic bromides would generate the same delocalized radical, but should exhibit different reactivity toward allylic substitution, with addition of a radical to the exomethylene double bond of 5 being faster than addition to allylic isomer 4 having a trisubstituted double bond. As an addition–fragmentation mechanism would be inconsistent with the formation of diene 8 containing two disubstituted double bonds in equal amounts from both bromide precursors, dimerization of allylic radicals is undoubtedly the major pathway. The somewhat higher yield

Scheme 2. Proposed Mechanism for the $\text{Ru}(\text{bpy})_3^{2+}$ -Catalyzed Generation of Allylic Radicals from the Corresponding Bromides and Possible Reaction Pathways



Scheme 3. Investigation of the Reaction Pathways Leading to the Observed Wurtz-Type Coupling Products^a



^aIsolated yield after silica gel chromatography (average of two experiments).

of diene **6** harboring two trisubstituted double bonds from allylic precursor **5** suggests that an addition–fragmentation pathway occurs to a limited extent, a conclusion that was exploited later in the reaction depicted in [Scheme 4](#).

After investigating the behavior of various substituted α -(bromomethyl)styrene derivatives in visible-light photocatalyzed coupling, allylic bromides with substitution other than an aryl moiety in the 2-position were tested ([Table 3](#)). Methyl 2-

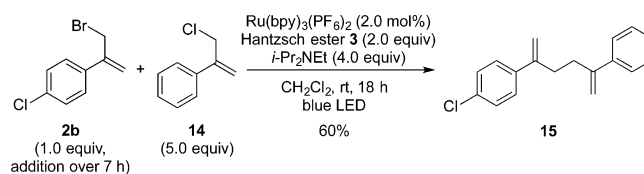
Table 3. Visible-Light-Mediated Photocatalytic Wurtz-Type Coupling of Various Allylic Bromides^a

Starting Material	Product (Yield in %) ^b
	 10 (57%)
	 12 (15%)
	ND

^aReaction conditions from [Table 2](#). ^bIsolated yield after silica gel chromatography. ND = not detected.

(bromomethyl)acrylate (**9**) gave diene product **10** in moderate yield. However, 2-bromoallyl bromide (**11**) delivered dibromodiene **12** in low yield only, whereas the attempted coupling of 2-methylallyl bromide (**13**) gave no 1,5-diene product. In all reactions summarized in [Table 3](#), no starting material was recovered after a reaction time of 18 h.¹⁹ Attempted extension of the method to 2-aryl-substituted allylic chlorides was briefly examined; however, only starting material was recovered in all cases. Even the more strongly reducing photocatalyst $\text{Ir}(\text{ppy})_3$ did not convert α -(chloromethyl)styrene to diene product **1**.

Scheme 4. Synthesis of Nonsymmetrically Substituted 2,5-Diaryl-1,5-dienes by Reaction via the Addition–Fragmentation Pathway



In order to access nonsymmetrically substituted 2,5-diaryl-1,5-dienes, reaction conditions that would favor the addition–fragmentation pathway over radical dimerization were examined. The investigations were conducted with 4-chloro- α -(bromomethyl)styrene (**2b**) as the radical precursor and α -(chloromethyl)styrene (**14**) as the acceptor ([Scheme 4](#)). Employing **14** in excess (5 equiv) under otherwise identical standard conditions led to a 1:1 mixture of dienes **1b** and **15** (67% overall yield). However, the addition of the bromide

precursor **2b** over 7 h using a syringe pump, which ensures a low concentration of allylic radicals in solution, and increasing the catalyst and reagent loading gave exclusively the unsymmetrical 2,5-diaryl-1,5-diene **15** in 60% yield.

CONCLUSION

2,5-Diarylhexa-1,5-dienes are formed in high yields at room temperature by reductive coupling of 2-aryllallyl bromides in the presence of 1 mol % of the commercially available photocatalyst Ru(bpy)₃(PF₆)₂, Hantzsch ester **3**, *i*-Pr₂NEt, and visible light. This attractive method avoids the use of stoichiometric metals and is believed to proceed largely via dimerization of photogenerated allylic radical intermediates. By slightly adapting the reaction conditions, the addition–fragmentation pathway is favored that grants access to nonsymmetrically substituted 2,5-diaryl-1,5-dienes.

EXPERIMENTAL SECTION

Materials and Methods. Unless stated otherwise, reactions were conducted in oven-dried glassware under an atmosphere of argon using anhydrous solvents (either freshly distilled or passed through activated alumina columns). For all photoredox-catalyzed coupling reactions, CH₂Cl₂ was sparged with argon for 5 min prior to use. All commercially obtained reagents were used as received. Ru(bpy)₃(PF₆)₂ and other photocatalysts were obtained from Sigma-Aldrich. 2-Methylallyl bromide (**13**), 2,3-dibromopropene (**11**), and TMS-Cl were distilled prior to use. The reaction components Hantzsch ester **3**,²⁰ methyl 2-(bromomethyl)acrylate (**9**),²¹ and α -(chloromethyl)styrene (**14**)²² were prepared according to literature procedures. Usually one representative coupling reaction and yield of the product is described in detail; isolated yields reported in the Results and Discussion section are the average yields obtained from duplicate experiments. Reaction temperatures were controlled using a temperature modulator, and unless stated otherwise, reactions were performed at room temperature (rt, approximately 23 °C). Thin-layer chromatography (TLC) was conducted with silica gel 60 F254 precoated plates (0.25 mm) and visualized by exposure to UV light (254 nm) or by anisaldehyde, ceric ammonium molybdate, iodine, and potassium permanganate staining. Silica gel 60 (particle size 0.040–0.063 mm) was used for flash column chromatography. ¹H NMR spectra were recorded at 500 or 600 MHz and are reported relative to deuterated solvent signals. Data for ¹H NMR spectra are reported as follows: chemical shift (δ ppm), multiplicity, coupling constant (Hz), and integration. ¹³C NMR spectra were recorded at 125 or 150 MHz. Data for ¹³C NMR spectra are reported in terms of chemical shift. IR spectra were recorded on an FT-IR spectrometer and are reported in terms of frequency of absorption (cm⁻¹). High-resolution mass spectra were obtained with an LCT spectrometer. Blue LEDs (30 cm, 1 W) were purchased from <http://www.creativelightings.com> (product code CL-FR5050-12WP-12V) and powered by 8 AA batteries.

Preparation of α -Methylstyrene Derivatives via Wittig Olefination (Scheme 1). General Procedure for the Wittig Olefination of Acetophenone Derivatives.¹⁵ A mixture of methyltriphenylphosphonium bromide (12.0–15.0 mmol, 4.32–5.40 g, 1.2–1.5 equiv) in dry THF (18.3 mL) under argon atmosphere was cooled to 0 °C in an ice bath. Then, *n*-BuLi (2.5 M solution in hexanes, 12.0–15.0 mmol, 4.8–6.0 mL, 1.2–1.5 equiv) was added slowly over 10–15 min under stirring. After the resulting orange mixture was maintained at 0 °C for 1 h, a solution of the acetophenone derivative (10.0 mmol, 1.0 equiv) in dry THF (7.0 mL) was added dropwise over 10–15 min at 0 °C. The reaction was allowed to warm up to room temperature, stirred overnight, and finally quenched with a saturated aqueous solution of sodium chloride (75 mL). The resulting mixture was extracted with pentane (3 \times 100 mL) or diethyl ether (3 \times 100 mL). The combined organic phases were washed with a saturated aqueous solution of sodium chloride, dried over sodium sulfate, and concentrated under reduced pressure. The crude residue was purified by silica gel chromatography.

4-Chloro- α -methylstyrene. Following the general procedure, the title compound was synthesized from 4-chloroacetophenone (10.0 mmol, 1.55 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 4-chloro- α -methylstyrene (9.51 mmol, 1.45 g, 95%) as a colorless oil. *R*_f 0.77 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.39 (d, *J* = 8.7, 2H), 7.29 (d, *J* = 8.7, 2H), 5.36 (s, 1H), 5.11–5.09 (m, 1H), 2.14–2.13 (m, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 142.3, 139.8, 133.3, 128.5, 126.9, 113.1, 21.9. Characterization data obtained matched those previously reported in the literature.²²

4-Bromo- α -methylstyrene. Following the general procedure, the title compound was synthesized from 4-bromoacetophenone (10.0 mmol, 1.99 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 4-bromo- α -methylstyrene (8.82 mmol, 1.74 g, 88%) as a colorless oil. *R*_f 0.76 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.45 (d, *J* = 8.7, 2H), 7.33 (d, *J* = 8.6, 2H), 5.36 (s, 1H), 5.11–5.10 (m, 1H), 2.13–2.12 (m, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 142.3, 140.2, 131.4, 127.3, 121.5, 113.2, 21.8. Characterization data obtained matched those previously reported in the literature.^{23,24}

3-Bromo- α -methylstyrene. Following the general procedure, the title compound was synthesized from 3-bromoacetophenone (10.0 mmol, 1.99 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 3-bromo- α -methylstyrene (9.53 mmol, 1.88 g, 95%) as a colorless oil. *R*_f 0.78 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.60 (t, *J* = 1.8, 1H), 7.41–7.38 (m, 2H), 7.20 (t, *J* = 7.9, 1H), 5.38 (s, 1H), 5.14–5.12 (m, 1H), 2.13 (s, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.5, 142.1, 130.4, 129.9, 128.8, 124.2, 122.6, 113.8, 21.8. Characterization data obtained matched those previously reported in the literature.²⁵

2-Fluoro- α -methylstyrene. Following the general procedure, the title compound was synthesized from 2-fluoroacetophenone (10.0 mmol, 1.38 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 2-fluoro- α -methylstyrene (8.84 mmol, 1.20 g, 88%) as a colorless oil. *R*_f 0.67 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.31 (dt, *J* = 7.8, 0.6, 1H), 7.25–7.21 (m, 1H), 7.10 (t, *J* = 7.5, 1H), 7.04 (dd, *J* = 11.1, 8.3, 1H), 5.24–5.22 (m, 2H), 2.16–2.14 (m, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 160.1 (d, *J* = 248.1), 140.3 (d, *J* = 0.8), 130.4 (d, *J* = 13.6), 129.5 (d, *J* = 4.4), 128.8 (d, *J* = 8.4), 124.0 (d, *J* = 3.5), 116.7 (d, *J* = 4.0), 116.0 (d, *J* = 23.0), 23.2 (d, *J* = 3.4); IR (thin film): 3083, 2974, 2924, 2855, 1633, 1573, 1489, 1448, 1216, 1092 cm⁻¹; HRMS-Cl (*m/z*) [M]⁺ calculated for C₉H₉F 136.0688, found 136.0687.

4-Methyl- α -methylstyrene. Following the general procedure, the title compound was synthesized from 4-methylacetophenone (10.0 mmol, 1.34 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 4-methyl- α -methylstyrene (8.76 mmol, 1.16 g, 88%) as a colorless oil. *R*_f 0.69 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.38 (d, *J* = 8.2, 2H), 7.15 (d, *J* = 7.9, 2H), 5.35 (s, 1H), 5.05–5.04 (m, 1H), 2.36 (s, 3H), 2.16–2.14 (m, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.2, 138.5, 137.3, 129.0, 125.5, 111.7, 22.0, 21.2. Characterization data obtained matched those previously reported in the literature.^{22,23}

3-Methyl- α -methylstyrene. Following the general procedure, the title compound was synthesized from 3-methylacetophenone (10.0 mmol, 1.34 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 3-methyl- α -methylstyrene (8.62 mmol, 1.14 g, 86%) as a colorless oil. *R*_f 0.69

(100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.30 (d, J = 8.9, 2H), 7.24 (t, J = 7.4, 1H), 7.11 (d, J = 7.6, 1H), 5.38–5.37 (m, 1H), 5.09–5.08 (m, 1H), 2.39 (s, 3H), 2.18–2.16 (m, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 143.5, 141.4, 137.8, 128.3, 128.2, 126.4, 122.8, 112.4, 22.0, 21.7. Characterization data obtained matched those previously reported in the literature.²⁶

2-Methyl- α -methylstyrene. Following the general procedure, the title compound was synthesized from 2-methylacetophenone (10.0 mmol, 1.34 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 2-methyl- α -methylstyrene (8.28 mmol, 1.09 g, 83%) as a colorless oil. R_f 0.50 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.19–7.11 (m, 4H), 5.21–5.19 (m, 1H), 4.86–4.85 (m, 1H), 2.33 (s, 3H), 2.05 (s, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 146.0, 144.0, 134.6, 130.2, 128.0, 126.9, 125.7, 114.8, 24.5, 20.0. Characterization data obtained matched those previously reported in the literature.²²

4-Methoxy- α -methylstyrene. Following the general procedure, the title compound was synthesized from 4-methoxyacetophenone (10.0 mmol, 1.50 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 4-methoxy- α -methylstyrene (8.61 mmol, 1.28 g, 86%) as a colorless oil. R_f 0.21 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.43 (d, J = 8.9, 2H), 6.88 (d, J = 8.9, 2H), 5.31–5.30 (m, 1H), 5.02–5.00 (m, 1H), 3.83 (s, 3H), 2.16–2.14 (m, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 159.2, 142.7, 133.9, 126.7, 113.7, 110.8, 55.4, 22.1. Characterization data obtained matched those previously reported in the literature.^{23,24,26}

3-Methoxy- α -methylstyrene. Following the general procedure, the title compound was synthesized from 3-methoxyacetophenone (10.0 mmol, 1.50 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 3-methoxy- α -methylstyrene (9.05 mmol, 1.34 g, 90%) as a colorless oil. R_f 0.25 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.26 (t, J = 8.0, 1H), 7.08 (d, J = 7.9, 1H), 7.02–7.01 (m, 1H), 6.83 (dd, J = 8.2, 2.3, 1H), 5.38 (s, 1H), 5.11–5.09 (m, 1H), 3.83 (s, 3H), 2.15 (s, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 159.6, 143.3, 142.9, 129.3, 118.2, 112.8, 112.7, 111.6, 55.4, 22.0. Characterization data obtained matched those previously reported in the literature.^{23,24}

2-Methoxy- α -methylstyrene. Following the general procedure, the title compound was synthesized from 2-methoxyacetophenone (10.0 mmol, 1.50 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided 2-methoxy- α -methylstyrene (7.98 mmol, 1.18 g, 80%) as a colorless oil. R_f 0.32 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.28–7.24 (m, 1H), 7.21 (dd, J = 7.5, 1.8, 1H), 6.94 (dt, J = 7.4, 1.0, 1H), 6.90 (d, J = 8.2, 1H), 5.18–5.16 (m, 1H), 5.09–5.07 (m, 1H), 3.85 (s, 3H), 2.15–2.13 (m, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 156.7, 144.5, 132.9, 129.5, 128.4, 120.6, 115.2, 110.9, 55.5, 23.3. Characterization data obtained matched those previously reported in the literature.²⁴

4-Cyano- α -methylstyrene. Following the general procedure, the title compound was synthesized from 4-acetylbenzonitrile (10.0 mmol, 1.45 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with diethyl ether and purification by silica gel chromatography (10% diethyl ether/hexanes) provided 4-cyano- α -methylstyrene (4.66 mmol, 667 mg, 47%) as a yellow oil. R_f 0.52 (10% ethyl acetate/hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.61 (d, J = 8.5, 2H), 7.54 (d, J = 8.5, 2H), 5.47 (s, 1H), 5.25–5.24 (m, 1H), 2.16 (s, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 145.8, 141.9, 132.2, 126.2, 119.1, 115.8, 111.0, 21.6. Characterization data obtained matched those previously reported in the literature.^{27,28}

3-Cyano- α -methylstyrene. Following the general procedure, the title compound was synthesized from 3-acetylbenzonitrile (10.0 mmol, 1.45 g, 1.0 equiv), methyltriphenylphosphonium bromide (12.0 mmol, 4.32 g, 1.2 equiv), and *n*-BuLi (2.5 M solution in hexanes, 12.0 mmol, 4.8 mL, 1.2 equiv). Extraction with diethyl ether and purification by silica gel chromatography (10% diethyl ether/hexanes) provided 3-cyano- α -methylstyrene (6.47 mmol, 926 mg, 65%) as a colorless oil. R_f 0.55 (10% ethyl acetate/hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.72 (s, 1H), 7.68 (dd, J = 8.0, 1.0, 1H), 7.55 (dd, J = 8.0, 1.0, 1H), 7.43 (t, J = 7.8, 1H), 5.42 (s, 1H), 5.21–5.19 (m, 1H), 2.16–2.14 (m, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 142.5, 141.4, 130.9, 129.9, 129.3, 129.2, 119.1, 114.8, 112.6, 21.6; IR (thin film): 3089, 2976, 2947, 2921, 2359, 2230, 1630, 1596, 1575, 1481, 1441, 1377, 1192 cm^{-1} ; HRMS-ESI (m/z) [$M + \text{Na}$] $^+$ calculated for $\text{C}_{10}\text{H}_9\text{NNa}$ 166.0633, found 166.0641.

But-1-en-2-ylbenzene. Following the general procedure, the title compound was synthesized from propiophenone (10.0 mmol, 1.34 g, 1.0 equiv), methyltriphenylphosphonium bromide (15.0 mmol, 5.40 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided but-1-en-2-ylbenzene (8.06 mmol, 1.07 g, 81%) as a colorless oil. R_f 0.75 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.42 (d, J = 7.6, 2H), 7.33 (t, J = 7.5, 2H), 7.26 (t, J = 7.3, 1H), 5.28 (s, 1H), 5.07–5.05 (m, 1H), 2.52 (q, J = 7.4, 2H), 1.11 (t, J = 7.4, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 150.2, 141.7, 128.4, 127.4, 126.1, 111.1, 28.2, 13.1. Characterization data obtained matched those previously reported in the literature.²⁹

But-2-en-2-ylbenzene. Following the general procedure, the title compound was synthesized from acetophenone (10.0 mmol, 1.20 g, 1.0 equiv), ethyltriphenylphosphonium bromide (15.0 mmol, 5.57 g, 1.5 equiv), and *n*-BuLi (2.5 M solution in hexanes, 15.0 mmol, 6.0 mL, 1.5 equiv). Extraction with pentane and purification by silica gel chromatography (100% pentane) provided a 1.7:1 mixture of *E/Z* isomers of but-2-en-2-ylbenzene (7.30 mmol, 966 mg, 73%) as a colorless oil. R_f 0.72 (100% hexanes); ^1H NMR (500 MHz, CDCl_3 , mixture of isomers = 0.6:0.4): δ 7.40–7.29 (m, 3.4H), 7.25–7.20 (m, 1.6H), 5.88 (dq, J = 6.9, 1.1, 0.6H), 5.58 (dq, J = 6.8, 1.1, 0.4H), 2.05 (s, 3.0H), 1.82 (d, J = 6.9, 1.8H), 1.63–1.59 (m, 1.2H); ^{13}C NMR (125 MHz, CDCl_3 , mixture of isomers): δ 144.2, 142.0, 136.9, 135.6, 128.3, 128.2, 128.1, 126.6, 126.5, 125.7, 122.6, 121.7, 25.5, 15.6, 15.0, 14.5. Characterization data obtained matched those previously reported in the literature.³⁰

Preparation of α -(Bromomethyl)styrene Derivatives via Allylic Bromination (Scheme 1). General Procedure for the Allylic Bromination of α -Methylstyrene Derivatives.¹⁶ To a mixture of the α -methylstyrene derivative (0.50 mmol, 1.0 equiv) and TMS-Cl (0.50–50 μmol , 1–6 μL , 0.1–10 mol %) in dry CH_2Cl_2 /THF (4:1, 1.5 mL) under an argon atmosphere were added NBS (0.60 mmol, 106 mg, 1.2 equiv) and Yb(OTf)₃ (0.50–50 μmol , 0.3–31 mg, 0.1–10 mol %) in one portion. After stirring for 1 h, the mixture was concentrated under reduced pressure. The resulting residue was filtered three times with pentane or diethyl ether, and the combined filtrates were concentrated under reduced pressure. The crude product mixture was then purified by silica gel chromatography.

α -(Bromomethyl)styrene (2a). Following the general procedure, 2a was synthesized from commercially available α -methylstyrene (0.50 mmol, 65 μL , 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided 2a (0.22 mmol, 43 mg, 43%) as a colorless oil. R_f 0.41 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.52–7.49 (m, 2H), 7.41–7.37 (m, 2H), 7.36–7.32 (m, 1H), 5.57 (s, 1H), 5.50 (s, 1H), 4.40 (s, 2H); ^{13}C NMR (125 MHz, CDCl_3): δ 144.4, 137.7, 128.7, 128.4, 126.2, 117.2, 34.4. Characterization data obtained for 2a matched those previously reported in the literature.²²

4-Chloro- α -(bromomethyl)styrene (2b). Following the general procedure, 2b was synthesized from 4-chloro- α -methylstyrene (0.50 mmol, 76 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1

mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2b** (0.26 mmol, 60 mg, 52%) as a colorless oil. R_f 0.40 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.43 (d, $J = 8.8$, 2H), 7.35 (d, $J = 8.8$, 2H), 5.54 (s, 1H), 5.50 (s, 1H), 4.35 (s, 2H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 143.3, 136.1, 134.3, 128.8, 127.6, 117.8, 34.0. Characterization data obtained for **2b** matched those previously reported in the literature.²²

4-Bromo- α -(bromomethyl)styrene (2c). Following the general procedure, **2c** was synthesized from 4-bromo- α -methylstyrene (0.50 mmol, 94 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2c** (0.25 mmol, 69 mg, 50%) as a colorless oil. R_f 0.41 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.50 (d, $J = 8.5$, 2H), 7.36 (d, $J = 8.6$, 2H), 5.55 (s, 1H), 5.51 (s, 1H), 4.35 (s, 2H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 143.4, 136.6, 131.8, 127.9, 122.5, 117.8, 33.9; IR (thin film): 2359, 2340, 1682, 1588, 1490, 1394, 1276, 1211, 1072, 1008 cm^{-1} ; HRMS-Cl (m/z) [M]⁺ calculated for $\text{C}_9\text{H}_8\text{Br}_2$ 273.8993, found 273.8994.

3-Bromo- α -(bromomethyl)styrene (2d). Following the general procedure, **2d** was synthesized from 3-bromo- α -methylstyrene (0.50 mmol, 94 mg, 1.0 equiv), TMS-Cl (25 μmol , 3 μL , 5 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (25 μmol , 16 mg, 5 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2d** (0.29 mmol, 81 mg, 59%) as a colorless oil. R_f 0.42 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.64–7.62 (m, 1H), 7.48–7.45 (m, 1H), 7.42–7.39 (m, 1H), 7.27–7.23 (m, 1H), 5.55 (s, 1H), 5.26 (s, 1H), 4.34 (s, 2H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 143.3, 139.9, 131.4, 130.2, 129.4, 124.9, 122.9, 118.5, 33.8; IR (thin film): 3062, 2969, 2359, 1623, 1591, 1556, 1476, 1395, 1297, 1210, 1070 cm^{-1} ; HRMS-Cl (m/z) [M]⁺ calculated for $\text{C}_9\text{H}_8\text{Br}_2$ 273.8993, found 273.8984.

2-Fluoro- α -(bromomethyl)styrene (2e). Following the general procedure, **2e** was synthesized from 2-fluoro- α -methylstyrene (0.50 mmol, 76 mg, 1.0 equiv), TMS-Cl (25 μmol , 3 μL , 5 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (25 μmol , 16 mg, 5 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2e** (0.24 mmol, 51 mg, 48%) as a colorless oil. R_f 0.46 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.37–7.29 (m, 2H), 7.15 (dt, $J = 7.5$, 0.8, 1H), 7.10–7.05 (m, 1H), 5.61 (s, 1H), 5.42 (s, 1H), 4.39 (s, 2H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 160.0 (d, $J = 247.9$), 141.4, 130.6 (d, $J = 4.0$), 130.0 (d, $J = 8.5$), 126.6 (d, $J = 13.8$), 124.3 (d, $J = 3.6$), 120.8 (d, $J = 2.7$), 116.0 (d, $J = 22.6$), 35.5 (d, $J = 5.3$). Characterization data obtained for **2e** matched those previously reported in the literature.³¹

4-Methyl- α -(bromomethyl)styrene (2f). Following the general procedure, **2f** was synthesized from 4-methyl- α -methylstyrene (0.50 mmol, 76 mg, 1.0 equiv), TMS-Cl (0.50 μmol , 1 μL , 0.1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (0.50 μmol , 0.3 mg, 0.1 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2f** (0.12 mmol, 25 mg, 24%) as a colorless oil. R_f 0.38 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.39 (d, $J = 8.2$, 2H), 7.19 (d, $J = 7.9$, 2H), 5.53 (s, 1H), 5.44 (s, 1H), 4.38 (s, 2H), 2.37 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 144.2, 138.3, 134.7, 129.4, 126.1, 116.5, 34.5, 21.3. Characterization data obtained for **2f** matched those previously reported in the literature.²²

3-Methyl- α -(bromomethyl)styrene (2g). Following the general procedure, **2g** was synthesized from 3-methyl- α -methylstyrene (0.50 mmol, 76 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2g** (0.22 mmol, 45 mg, 43%) as a colorless oil. R_f 0.35 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.31–7.25 (m, 3H), 7.16 (d, $J = 6.6$, 1H), 5.54 (s, 1H), 5.48 (s, 1H), 4.38 (s, 2H), 2.39 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 144.5, 138.2, 137.7, 129.2, 128.5, 127.0, 123.3, 117.2, 34.5, 21.7; IR (thin film): 3031, 2921, 2361, 2341, 1602, 1489, 1442, 1278, 1210 cm^{-1} ; HRMS-Cl (m/z) [M]⁺ calculated for $\text{C}_{10}\text{H}_{11}\text{Br}$ 210.0044, found 210.0046.

2-Methyl- α -(bromomethyl)styrene (2h). Following the general procedure, **2h** was synthesized from 2-methyl- α -methylstyrene (0.50 mmol, 76 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **2h** (0.28 mmol, 58 mg, 55%) as a colorless oil. R_f 0.52 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.25–7.16 (m, 3H), 5.64 (s, 1H), 5.15–5.14 (m, 1H), 4.25 (s, 2H), 2.34 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 145.1, 139.4, 135.4, 130.4, 128.9, 127.9, 125.7, 119.6, 37.0, 19.9. Characterization data obtained for **2h** matched those previously reported in the literature.²²

4-Methoxy- α -(bromomethyl)styrene (2i). A mixture of 4-methoxy- α -methylstyrene (2.00 mmol, 296 mg, 1.0 equiv), NBS (1.23 mmol, 220 mg, 0.62 equiv), and benzoyl peroxide (0.10 mmol, 24 mg, 5 mol %) in CCl_4 (18 mL) was refluxed at 78 °C under an argon atmosphere. After 3 h, a second portion of NBS (1.23 mmol, 220 mg, 0.62 equiv) and benzoyl peroxide (0.10 mmol, 24 mg, 5 mol %) was added, and the reaction mixture was refluxed for three additional hours. After cooling to room temperature, the resulting precipitate was removed by filtration and the filtrate was concentrated under reduced pressure. Purification of the crude product by silica gel chromatography (2% diethyl ether/hexanes) provided **2i** (0.91 mmol, 207 mg, 46%) as a colorless oil. R_f 0.37 (5% ethyl acetate/hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.45 (d, $J = 8.8$, 2H), 6.91 (d, $J = 8.8$, 2H), 5.48 (s, 1H), 5.40 (s, 1H), 4.37 (s, 2H), 3.83 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 159.8, 143.7, 130.1, 127.7, 115.6, 114.0, 55.4, 34.6. Characterization data obtained for **2i** matched those previously reported in the literature.³²

3-Methoxy- α -(bromomethyl)styrene (2j). Following the general procedure, **2j** was synthesized from 3-methoxy- α -methylstyrene (0.50 mmol, 74 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 1 μL , 1 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 3 mg, 1 mol %). Filtration with diethyl ether and purification by silica gel chromatography (2% diethyl ether/hexanes) provided **2j** (0.22 mmol, 49 mg, 43%) as a colorless oil. R_f 0.53 (5% ethyl acetate/hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.30 (t, $J = 8.0$, 1H), 7.08 (d, $J = 7.7$, 1H), 7.04–7.02 (m, 1H), 6.88 (dd, $J = 8.2$, 2.5, 1H), 5.56 (s, 1H), 5.49 (s, 1H), 4.37 (s, 2H), 3.84 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 159.8, 144.3, 139.3, 129.6, 118.7, 117.6, 113.6, 112.3, 55.4, 34.3; IR (thin film): 2957, 2833, 2364, 1599, 1577, 1490, 1453, 1427, 1323, 1288, 1237, 1211, 1046 cm^{-1} ; HRMS-Cl (m/z) [M]⁺ calculated for $\text{C}_{10}\text{H}_{11}\text{BrO}$ 225.9993, found 225.9983.

2-Methoxy- α -(bromomethyl)styrene (2k). A mixture of 2-methoxy- α -methylstyrene (2.00 mmol, 296 mg, 1.0 equiv), NBS (1.23 mmol, 220 mg, 0.62 equiv), and benzoyl peroxide (0.10 mmol, 24 mg, 5 mol %) in CCl_4 (18 mL) was refluxed at 78 °C under an argon atmosphere. After 3 h, a second portion of NBS (1.23 mmol, 220 mg, 0.62 equiv) and benzoyl peroxide (0.10 mmol, 24 mg, 5 mol %) was added, and the reaction mixture was refluxed for three additional hours. After cooling to room temperature, the resulting precipitate was removed by filtration and the filtrate was concentrated under reduced pressure. Purification of the crude product by silica gel chromatography (2% diethyl ether/hexanes) provided **2k** (0.90 mmol, 204 mg, 45%) as a colorless oil. R_f 0.50 (5% ethyl acetate/hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.36–7.32 (m, 1H), 7.27 (dd, $J = 7.5$, 1.8, 1H), 6.99 (dt, $J = 7.4$, 0.8, 1H), 6.92 (d, $J = 8.4$, 1H), 5.53 (s, 1H), 5.29–5.28 (m, 1H), 4.49 (s, 2H), 3.85 (s, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 156.6, 145.1, 131.0, 129.5, 128.3, 120.7, 119.3, 110.7, 55.5, 36.1; IR (thin film): 2936, 2834, 1626, 1598, 1490, 1461, 1434, 1242, 1211, 1026 cm^{-1} ; HRMS-Cl (m/z) [$\text{M} + \text{NH}_4$]⁺ calculated for $\text{C}_{10}\text{H}_{11}\text{BrONH}_4$ 244.0337, found 244.0335.

4-Cyano- α -(bromomethyl)styrene (2l). Following the general procedure, **2l** was synthesized from 4-cyano- α -methylstyrene (0.50 mmol, 72 mg, 1.0 equiv), TMS-Cl (5.0 μmol , 6 μL , 10 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and Yb(OTf)₃ (5.0 μmol , 31 mg, 10 mol %). Filtration with diethyl ether and purification by silica gel chromatography (10% diethyl ether/hexanes) provided **2l** (0.26 mmol, 58 mg, 52%) as a colorless solid. R_f 0.34 (10% ethyl acetate/hexanes); mp: 58–59 °C; $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.67 (d, $J = 8.2$, 2H),

7.59 (d, $J = 8.2$, 2H), 5.65 (s, 1H), 5.63 (s, 1H), 4.36 (s, 2H); ^{13}C NMR (125 MHz, CDCl_3): δ 143.1, 142.2, 132.5, 127.0, 120.0, 118.7, 112.0, 33.2; IR (thin film): 2921, 2360, 2227, 1606, 1506, 1447, 1403, 1212 cm^{-1} ; HRMS-ESI (m/z) [$\text{M} + \text{Na}$] $^+$ calculated for $\text{C}_{10}\text{H}_8\text{NBrNa}$ 243.9738, found 243.9729.

3-Cyano- α -(bromomethyl)styrene (2m). Following the general procedure, **2m** was synthesized from 3-cyano- α -methylstyrene (0.50 mmol, 72 mg, 1.0 equiv), TMS-Cl (50 μmol , 6 μL , 10 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and $\text{Yb}(\text{OTf})_3$ (50 μmol , 31 mg, 10 mol %). Filtration with diethyl ether and purification by silica gel chromatography (10% diethyl ether/hexanes) provided **2m** (0.21 mmol, 47 mg, 43%) as a colorless solid. R_f 0.33 (10% ethyl acetate/hexanes); mp: 61–63 $^\circ\text{C}$; ^1H NMR (500 MHz, CDCl_3): δ 7.76 (s, 1H), 7.71 (d, $J = 8.0$, 1H), 7.62 (d, $J = 7.7$, 1H), 7.49 (t, $J = 7.8$, 1H), 5.59 (s, 2H), 4.35 (s, 2H); ^{13}C NMR (125 MHz, CDCl_3): δ 142.6, 139.0, 131.7, 130.6, 130.0, 129.5, 119.3, 118.7, 112.9, 33.3; IR (thin film): 2920, 2360, 2340, 2229, 1573, 1481, 1277, 1260, 1212 cm^{-1} ; HRMS-ESI (m/z) [$\text{M} + \text{Na}$] $^+$ calculated for $\text{C}_{10}\text{H}_8\text{NBrNa}$ 243.9738, found 243.9746.

(Z)-(1-Bromobut-2-en-2-yl)benzene (4). Following the general procedure, **4** was synthesized from but-1-en-2-ylbenzene (0.50 mmol, 66 mg, 1.0 equiv), TMS-Cl (12.5 μmol , 2 μL , 2.5 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and $\text{Yb}(\text{OTf})_3$ (12.5 μmol , 8 mg, 2.5 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **4** (0.30 mmol, 62 mg, 59%) as a colorless oil. R_f 0.32 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.44 (d, $J = 7.9$, 2H), 7.35 (t, $J = 7.6$, 2H), 7.29 (d, $J = 7.4$, 1H), 6.09 (q, $J = 7.1$, 1H), 4.40 (s, 2H), 1.92 (d, $J = 7.1$, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 140.6, 137.0, 129.5, 128.6, 127.5, 125.9, 29.3, 14.6; IR (thin film): 3027, 2977, 2918, 1600, 1493, 1446, 1207 cm^{-1} ; HRMS-Cl (m/z) [$\text{M} + \text{NH}_4$] $^+$ calculated for $\text{C}_{10}\text{H}_{11}\text{BrNH}_4$ 228.0388, found 228.0383.

(3-Bromobut-1-en-2-yl)benzene (5). Following the general procedure, **5** was synthesized from but-2-en-2-ylbenzene (0.50 mmol, 66 mg, 1.0 equiv), TMS-Cl (12.5 μmol , 2 μL , 2.5 mol %), NBS (0.60 mmol, 106 mg, 1.2 equiv), and $\text{Yb}(\text{OTf})_3$ (12.5 μmol , 8 mg, 2.5 mol %). Filtration with pentane and purification by silica gel chromatography (100% pentane) provided **5** (0.35 mmol, 74 mg, 70%) as a colorless oil. R_f 0.38 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.45 (d, $J = 7.0$, 2H), 7.39–7.31 (m, 3H), 5.53 (s, 1H), 5.40 (s, 1H), 5.12 (q, $J = 6.7$, 1H), 1.94 (d, $J = 6.8$, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 150.4, 139.9, 128.5, 128.1, 127.0, 115.3, 48.9, 24.9; IR (thin film): 3056, 2975, 2925, 2362, 1624, 1574, 1494, 1443, 1374, 1172, 1071 cm^{-1} ; HRMS-Cl (m/z) [M] $^+$ calculated for $\text{C}_{10}\text{H}_{11}\text{Br}$ 210.0044, found 210.0034.

Visible-Light Photoredox Catalyzed Wurtz-Type Coupling Reactions (Tables 1–3 and Scheme 3). General Procedure for the Wurtz-Type Coupling Reaction of Substituted Allylic Bromides. A 1 dram vial was charged with $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), and a magnetic stir bar under argon. After sequential addition of CH_2Cl_2 (0.5 mL, sparged with argon for 5 min), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and the substituted allylic bromide (0.20 mmol, 1.0 equiv), the vial was capped and placed in the center of a 30 cm loop of blue LEDs. After stirring for 18 h, the reaction mixture was concentrated under reduced pressure. The crude product was then purified by silica gel chromatography.

2,5-Diphenylhexa-1,5-diene (1a). Following the general procedure, **1a** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and α -(bromomethyl)styrene (**2a**, 0.20 mmol, 39 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1a** (81 μmol , 19 mg, 81%) as a colorless solid.

Under the same conditions, coupling product **1a** (94 mg, 80%) was obtained from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (0.01 mmol, 8.5 mg, 1 mol %), Hantzsch ester **3** (1.00 mmol, 255 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (2.00 mmol, 0.32 mL, 2.0 equiv), and α -(bromomethyl)styrene (**2a**, 1.00 mmol, 197 mg, 1.0 equiv) in CH_2Cl_2 (2.5 mL). In the same fashion, product **1a** (184 mg, 78%) was obtained from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (0.02

mmol, 17 mg, 1 mol %), Hantzsch ester **3** (2.00 mmol, 510 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (4.00 mmol, 0.63 mL, 2.0 equiv), and α -(bromomethyl)styrene (**2a**, 2.00 mmol, 394 mg, 1.0 equiv) in CH_2Cl_2 (5 mL).

R_f 0.37 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.42 (d, $J = 7.5$, 4H), 7.35 (t, $J = 7.4$, 4H), 7.32–7.28 (m, 2H), 5.31 (d, $J = 1.2$, 2H), 5.07 (s, 2H), 2.68 (s, 4H); ^{13}C NMR (125 MHz, CDCl_3): δ 148.1, 141.2, 128.4, 127.5, 126.3, 112.7, 34.4. Characterization data obtained for **1a** matched those previously reported in the literature.³³

2,5-Bis(4-chlorophenyl)hexa-1,5-diene (1b). Following the general procedure, **1b** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and 4-chloro- α -(bromomethyl)styrene (**2b**, 0.20 mmol, 46 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1b** (81 μmol , 25 mg, 81%) as a colorless solid. R_f 0.44 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.29 (s, 8H), 5.26 (s, 2H), 5.03 (s, 2H), 2.59 (s, 4H); ^{13}C NMR (125 MHz, CDCl_3): δ 146.8, 139.5, 133.3, 128.6, 127.6, 113.5, 34.2. Characterization data obtained for **1b** matched those previously reported in the literature.^{14k}

2,5-Bis(4-bromophenyl)hexa-1,5-diene (1c). Following the general procedure, **1c** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and 4-bromo- α -(bromomethyl)styrene (**2c**, 0.20 mmol, 55 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1c** (83 μmol , 32 mg, 83%) as a colorless solid. R_f 0.41 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.45 (d, $J = 8.4$, 4H), 7.23 (d, $J = 8.4$, 4H), 5.26 (s, 2H), 5.03 (s, 2H), 2.59 (s, 4H); ^{13}C NMR (125 MHz, CDCl_3): δ 146.8, 140.0, 131.5, 127.9, 121.5, 113.6, 34.1. Characterization data obtained for **1c** matched those previously reported in the literature.^{14k}

2,5-Bis(3-bromophenyl)hexa-1,5-diene (1d). Following the general procedure, **1d** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and 3-bromo- α -(bromomethyl)styrene (**2d**, 0.20 mmol, 55 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1d** (87 μmol , 34 mg, 87%) as a colorless oil.

Under the same conditions, coupling product **1d** (154 mg, 79%) was obtained from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (0.01 mmol, 8.5 mg, 1 mol %), Hantzsch ester **3** (1.00 mmol, 255 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (2.00 mmol, 0.32 mL, 2.0 equiv), and 3-bromo- α -(bromomethyl)styrene (**2d**, 1.00 mmol, 275 mg, 1.0 equiv) in CH_2Cl_2 (2.5 mL).

R_f 0.50 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.50 (t, $J = 1.8$, 2H), 7.42–7.39 (m, 2H), 7.30–7.26 (m, 2H), 7.20 (t, $J = 7.8$, 2H), 5.28 (d, $J = 1.0$, 2H), 5.07 (s, 2H), 2.59 (s, 4H); ^{13}C NMR (125 MHz, CDCl_3): δ 146.6, 143.3, 130.5, 130.0, 129.4, 124.9, 122.7, 114.2, 34.1; IR (thin film): 3081, 2942, 2859, 1806, 1626, 1589, 1556, 1473, 1407, 1291, 1068 cm^{-1} ; HRMS-Cl (m/z) [$\text{M} + \text{NH}_4$] $^+$ calculated for $\text{C}_{18}\text{H}_{16}\text{Br}_2\text{NH}_4$ 407.9962, found 407.9958.

2,5-Bis(2-fluorophenyl)hexa-1,5-diene (1e). Following the general procedure, **1e** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and 2-fluoro- α -(bromomethyl)styrene (**2e**, 0.20 mmol, 43 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1e** (66 μmol , 18 mg, 66%) as a colorless oil. R_f 0.45 (100% hexanes); ^1H NMR (500 MHz, CDCl_3): δ 7.25–7.18 (m, 4H), 7.08 (dt, $J = 7.5$, 1.1, 2H), 7.02 (ddd, $J = 10.7$, 8.3, 0.9, 2H), 5.17 (s, 2H), 5.15 (s, 2H), 2.58 (s, 4H); ^{13}C NMR (125 MHz, CDCl_3): δ 160.0 (d, $J = 247.3$), 144.4, 130.3 (d, $J = 4.4$), 129.8 (d, $J = 14.3$), 128.9 (d, $J = 8.4$), 124.0 (d, $J = 3.5$), 116.3 (d, $J = 2.3$), 115.8 (d, $J = 22.9$), 35.0 (d, $J = 3.2$); IR (thin film): 3082, 2933, 2860, 1799, 1631, 1573, 1487, 1447, 1214, 1090, 1033 cm^{-1} ; HRMS-Cl (m/z) [$\text{M} + \text{H}$] $^+$ calculated for $\text{C}_{18}\text{H}_{16}\text{F}_2\text{H}$ 271.1298, found 271.1292.

2,5-Bis(4-methylphenyl)hexa-1,5-diene (1f). Following the general procedure, **1f** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μmol , 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), $i\text{-Pr}_2\text{NEt}$ (0.40 mmol, 63 μL , 2.0 equiv), and 4-methyl- α -(bromomethyl)styrene (**2f**, 0.20 mmol, 42 mg, 1.0 equiv) in CH_2Cl_2 (1.0 mL). Purification by

silica gel chromatography (100% pentane) provided **1f** (82 μ mol, 22 mg, 82%) as a colorless solid. R_f 0.19 (100% hexanes); mp: 56–57 °C; $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.29 (d, J = 8.1, 4H), 7.14 (d, J = 7.9, 4H), 5.25 (d, J = 1.3, 2H), 5.00 (s, 2H), 2.62 (s, 4H), 2.35 (s, 6H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 148.0, 138.3, 137.2, 129.1, 126.1, 111.8, 34.5, 21.3; IR (thin film): 3082, 3023, 2921, 2861, 1624, 1512, 1453 cm^{-1} ; HRMS-Cl (m/z) [$M + \text{H}$] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{H}$ 263.1800, found 263.1804.

2,5-Bis(3-methylphenyl)hexa-1,5-diene (1g). Following the general procedure, **1g** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 3-methyl- α -(bromomethyl)styrene (**2g**, 0.20 mmol, 42 mg, 1.0 equiv) in CH_2Cl_2 (1.0 mL). Purification by silica gel chromatography (100% pentane) provided **1g** (88 μ mol, 23 mg, 88%) as a colorless oil. R_f 0.36 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.24–7.18 (m, 6H), 7.10 (d, J = 6.9, 2H), 5.27 (d, J = 1.4, 2H), 5.04 (s, 2H), 2.63 (s, 4H), 2.36 (s, 6H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 148.3, 141.2, 137.9, 128.3, 128.2, 127.0, 123.4, 112.5, 34.5, 21.7; IR (thin film): 3035, 2919, 2859, 2360, 1787, 1626, 1600, 1575, 1486, 1452 cm^{-1} ; HRMS-Cl (m/z) [$M + \text{H}$] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{H}$ 263.1800, found 263.1792.

2,5-Bis(2-methylphenyl)hexa-1,5-diene (1h). Following the general procedure, **1h** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 2-methyl- α -(bromomethyl)styrene (**2h**, 0.20 mmol, 42 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **1h** (72 μ mol, 19 mg, 72%) as a colorless oil.

Under the same conditions, coupling product **1h** (90 mg, 69%) was obtained from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (0.01 mmol, 8.5 mg, 1 mol %), Hantzsch ester **3** (1.00 mmol, 255 mg, 1.0 equiv), *i*-Pr₂NEt (2.00 mmol, 0.32 mL, 2.0 equiv), and 2-methyl- α -(bromomethyl)styrene (**2h**, 1.00 mmol, 210 mg, 1.0 equiv) in CH_2Cl_2 (2.5 mL).

R_f 0.39 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.18–7.11 (m, 6H), 7.06 (d, J = 6.9, 2H), 5.20 (s, 2H), 4.89 (d, J = 1.9, 2H), 2.45 (s, 4H), 2.26 (s, 6H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 149.6, 143.0, 134.9, 130.2, 128.5, 126.9, 125.5, 114.1, 35.9, 20.0; IR (thin film): 3070, 3016, 2924, 2360, 1634, 1487, 1452, 1211, 1045 cm^{-1} ; HRMS-Cl (m/z) [$M + \text{H}$] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{H}$ 263.1800, found 263.1804.

2,5-Bis(4-methoxyphenyl)hexa-1,5-diene (1i). Following the general procedure, **1i** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 4-methoxy- α -(bromomethyl)styrene (**2i**, 0.20 mmol, 45 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (5% diethyl ether/hexanes) provided **1i** (63 μ mol, 19 mg, 63%) as a colorless solid. R_f 0.45 (10% ethyl acetate/hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.35 (d, J = 8.5, 4H), 6.87 (d, J = 8.4, 4H), 5.22 (s, 2H), 4.97 (s, 2H), 3.82 (s, 6H), 2.62 (s, 4H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 159.2, 147.5, 133.6, 127.3, 113.7, 111.1, 55.4, 34.5. Characterization data obtained for **1i** matched those previously reported in the literature.^{14b}

2,5-Bis(3-methoxyphenyl)hexa-1,5-diene (1j). Following the general procedure, **1j** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 3-methoxy- α -(bromomethyl)styrene (**2j**, 0.20 mmol, 45 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (5% diethyl ether/hexanes) provided **1j** (74 μ mol, 22 mg, 74%) as a colorless oil. R_f 0.50 (10% ethyl acetate/hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.27–7.23 (m, 2H), 6.99 (d, J = 7.8, 2H), 6.94–6.92 (m, 2H), 6.83 (dd, J = 8.2, 2.5, 2H), 5.29 (d, J = 1.2, 1H), 5.06 (s, 2H), 3.82 (s, 6H), 2.64 (s, 4H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 159.7, 148.1, 142.8, 129.4, 118.6, 112.9, 112.8, 112.2, 55.3, 34.5; IR (thin film): 3078, 2934, 2833, 1602, 1575, 1487, 1460, 1427, 1286, 1231, 1047 cm^{-1} ; HRMS-Cl (m/z) [$M + \text{H}$] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{O}_2\text{H}$ 295.1698, found 295.1700.

2,5-Bis(2-methoxyphenyl)hexa-1,5-diene (1k). Following the general procedure, **1k** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (3.0 μ mol, 3 mg, 1.5 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0

equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 2-methoxy- α -(bromomethyl)styrene (**2k**, 0.20 mmol, 45 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). The reaction time was extended from 18 to 42 h. Purification by silica gel chromatography (5% diethyl ether/hexanes) provided **1k** (69 μ mol, 20 mg, 69%) as a colorless solid. R_f 0.44 (5% ethyl acetate/hexanes); mp: 98–100 °C; $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.26–7.22 (m, 2H), 7.10 (dd, J = 7.4, 1.7, 2H), 6.90 (dt, J = 7.4, 1.0, 2H), 6.85 (d, J = 8.2, 2H), 5.12 (s, 2H), 5.00 (d, J = 2.0, 2H), 3.75 (s, 6H), 2.55 (s, 4H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 156.6, 148.9, 132.2, 130.4, 128.4, 120.5, 114.2, 110.7, 55.5, 35.0; IR (thin film): 3072, 2944, 2837, 1633, 1597, 1489, 1458, 1435, 1239, 1024 cm^{-1} ; HRMS-Cl (m/z) [M] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{O}_2$ 294.1620, found 294.1625.

2,5-Bis(4-cyanophenyl)hexa-1,5-diene (1l). Following the general procedure, **1l** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 4-cyano- α -(bromomethyl)styrene (**2l**, 0.20 mmol, 44 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL). Purification by silica gel chromatography (10% \rightarrow 20% diethyl ether/hexanes) provided **1l** (73 μ mol, 21 mg, 73%) as a colorless solid. R_f 0.18 (10% ethyl acetate/hexanes); mp: 115–116 °C; $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.62 (d, J = 8.4, 4H), 7.44 (d, J = 8.4, 4H), 5.37 (s, 2H), 5.15 (s, 2H), 2.63 (s, 4H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 146.1, 145.5, 132.4, 126.9, 118.9, 116.1, 111.2, 33.6; IR (thin film): 3065, 2917, 2852, 2360, 2341, 2226, 1623, 1604, 1504, 1401, 1128 cm^{-1} ; HRMS-ESI (m/z) [$M + \text{Na}$] $^+$ calculated for $\text{C}_{20}\text{H}_{16}\text{N}_2\text{Na}$ 307.1211, found 307.1213.

2,5-Bis(3-cyanophenyl)hexa-1,5-diene (1m). Following the general procedure, **1m** was synthesized from $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and 3-cyano- α -(bromomethyl)styrene (**2m**, 0.20 mmol, 44 mg, 1.0 equiv) in CH_2Cl_2 (1.0 mL). Purification by silica gel chromatography (20% diethyl ether/hexanes) provided **1m** (74 μ mol, 21 mg, 74%) as a colorless solid. R_f 0.51 (20% ethyl acetate/hexanes); mp: 64–65 °C; $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.61–7.59 (m, 2H), 7.58–7.55 (m, 4H), 7.46–7.42 (m, 2H), 5.32 (s, 2H), 5.12 (s, 2H), 2.61 (s, 4H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 145.6, 142.2, 131.1, 130.6, 129.9, 129.4, 118.9, 115.4, 112.7, 33.6; IR (thin film): 3082, 2921, 2851, 2360, 2229, 1627, 1574, 1480 cm^{-1} ; HRMS-ESI (m/z) [$M + \text{Na}$] $^+$ calculated for $\text{C}_{20}\text{H}_{16}\text{N}_2\text{Na}$ 307.1211, found 307.1224.

Wurtz-Type Coupling of (Z)-(1-Bromobut-2-en-2-yl)benzene (4). Following the general procedure, the reaction of $\text{Ru}(\text{bpy})_3(\text{PF}_6)_2$ (2.0 μ mol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μ L, 2.0 equiv), and (Z)-(1-bromobut-2-en-2-yl)benzene (**4**, 0.20 mmol, 42 mg, 1.0 equiv) in CH_2Cl_2 (0.5 mL) gave a crude product. Purification by silica gel chromatography (100% pentane) provided a mixture of the three regioisomers **6** (36 μ mol, 10 mg, 36%), **7** (43 μ mol, 11 mg, 43%), and **8** (15 μ mol, 4 mg, 15%). Analytical samples were obtained by an automated flash chromatography system.

Data for **6**: colorless solid; R_f 0.45 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.31–7.28 (m, 8H), 7.24–7.20 (m, 2H), 5.72 (q, J = 6.9, 2H), 2.54 (s, 4H), 1.67 (d, J = 6.9, 6H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 143.1, 140.5, 128.3, 126.6, 126.4, 123.4, 28.1, 14.2. Characterization data obtained for **6** matched those previously reported in the literature.³⁴

Data for **7**: colorless oil; R_f 0.50 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.30–7.19 (m, 10H), 5.75 (q, J = 6.8, 1H), 5.17 (d, J = 0.9, 1H), 5.08–5.07 (m, 1H), 2.70–2.64 (m, 2H), 2.50–2.45 (m, 1H), 1.77 (d, J = 6.9, 3H), 1.07 (d, J = 6.9, 3H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 155.1, 143.3, 142.9, 140.1, 128.3, 128.2, 127.3, 126.9, 126.7, 126.6, 124.6, 111.0, 36.4, 35.7, 18.6, 14.6; IR (thin film): 3022, 3961, 2925, 2360, 1493, 1442, 1371, 1028 cm^{-1} ; HRMS-Cl (m/z) [$M + \text{H}$] $^+$ calculated for $\text{C}_{20}\text{H}_{22}\text{H}$ 263.1800, found 263.1797.

Data for **8** (1:1 mixture of diastereomers): colorless oil; R_f 0.53 (100% hexanes); $^1\text{H NMR}$ (500 MHz, CDCl_3 , mixture of diastereomers): δ 7.34–7.24 (m, 16H), 7.20–7.16 (m, 4H), 5.20–5.19 (m, 2H), 5.19 (s, 2H), 5.06 (s, 2H), 5.01–5.00 (m, 2H), 2.79–2.72 (m, 4H), 1.18 (d, J = 6.2, 6H), 0.94 (d, J = 6.8, 6H).

Characterization data obtained for **8** matched those previously reported in the literature.³⁴

Wurtz-Type Coupling of (3-Bromobut-1-en-2-yl)benzene (5). Following the general procedure, the reaction of Ru(bpy)₃(PF₆)₂ (2.0 μmol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μL, 2.0 equiv), and (3-bromobut-1-en-2-yl)benzene (**5**, 0.20 mmol, 42 mg, 1.0 equiv) in CH₂Cl₂ (0.5 mL) gave a crude product. Purification by silica gel chromatography (100% pentane) provided the three regioisomers **6** (45 μmol, 12 mg, 45%), **7** (39 μmol, 10 mg, 39%), and **8** (14 μmol, 4 mg, 14%). Characterization data obtained for **6–8** matched those reported above.

Dimethyl 2,5-Dimethylenehexanedioate (10). Following the general procedure, **10** was synthesized from Ru(bpy)₃(PF₆)₂ (2.0 μmol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μL, 2.0 equiv), and methyl 2-(bromomethyl)acrylate² (**9**, 0.20 mmol, 36 mg, 1.0 equiv) in CH₂Cl₂ (0.5 mL). Purification by silica gel chromatography (10% diethyl ether/hexanes) provided **10** (57 μmol, 11 mg, 57%) as a colorless oil. *R*_f 0.37 (10% ethyl acetate/hexanes); ¹H NMR (500 MHz, CDCl₃): δ 6.17 (d, *J* = 1.3, 2H), 5.54 (s, 2H), 3.76 (s, 6H), 2.50 (s, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 167.6, 139.6, 125.7, 52.0, 31.0; IR (thin film): 2923, 2851, 2361, 1721, 1632, 1439, 1203, 1141 cm⁻¹; HRMS-CI (*m/z*) [M + H]⁺ calculated for C₁₀H₁₄O₄H 199.0970, found 199.0976.

2,5-Dibromohexa-1,5-diene (12). Following the general procedure, **12** was synthesized from Ru(bpy)₃(PF₆)₂ (2.0 μmol, 2 mg, 1 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 1.0 equiv), *i*-Pr₂NEt (0.40 mmol, 63 μL, 2.0 equiv), and 2,3-dibromopropene (**11**, 0.20 mmol, 40 mg, 1.0 equiv) in CH₂Cl₂ (0.5 mL). Purification by silica gel chromatography (100% pentane) provided **12** (15 μmol, 4 mg, 15%) as a colorless oil. *R*_f 0.78 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 5.64 (d, *J* = 1.7, 2H), 5.45 (d, *J* = 1.8, 2H), 2.68 (s, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 132.0, 118.3, 39.9. Characterization data obtained for **12** matched those previously reported in the literature.³⁵

Visible-Light Photoredox Catalyzed Coupling to Activated Olefins (Scheme 4). **2-(4-Chlorophenyl)-5-phenylhexa-1,5-diene (15).** A 1 dram vial was charged with Ru(bpy)₃(PF₆)₂ (2.0 μmol, 2 mg, 2 mol %), Hantzsch ester **3** (0.20 mmol, 51 mg, 2.0 equiv), and a magnetic stir bar under argon. After sequential addition of CH₂Cl₂ (0.5 mL, sparged with argon for 5 min), *i*-Pr₂NEt (0.40 mmol, 63 μL, 4.0 equiv), and α -(chloromethyl)styrene (**14**, 0.50 mmol, 76 mg, 5.0 equiv), the vial was placed in the center of a 30 cm loop of blue LEDs. A solution of 4-chloro- α -(bromomethyl)styrene (**2b**, 0.10 mmol, 23 mg, 1.0 equiv) in CH₂Cl₂ (1.0 mL, sparged with argon for 5 min) was added over a time period of 7 h with a syringe pump. After additional stirring for 11 h, the reaction mixture was concentrated under reduced pressure. Purification by silica gel chromatography (100% pentane) provided **15** (60 μmol, 16 mg, 60%) as a colorless oil. *R*_f 0.31 (100% hexanes); ¹H NMR (500 MHz, CDCl₃): δ 7.39–7.27 (m, 9H), 5.27 (d, *J* = 1.1, 1H), 5.26 (d, *J* = 1.1, 1H), 5.05 (d, *J* = 0.9, 1H), 5.03 (d, *J* = 0.9, 1H), 2.66–2.58 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 148.0, 147.0, 141.1, 139.6, 133.3, 128.6, 128.5, 128.4, 127.6, 127.5, 126.3, 113.3, 112.9, 34.3, 34.2; IR (thin film): 3081, 2939, 2361, 1626, 1492, 1443, 1394, 1095, 1012 cm⁻¹; HRMS-CI (*m/z*) [M + H]⁺ calculated for C₁₈H₁₇ClH 269.1097, found 269.1090.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b01962.

Copies of ¹H and ¹³C NMR spectra of compounds (PDF)

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Notes

The authors declare no competing financial interest.

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